

**GLUTATHIONE SYNTHESIS IN FILARIAL WORMS: AN ATTRACTIVE  
TARGET FOR THE DESIGN AND SYNTHESIS OF NEW ANTIFILARIALS**

Kavita Arora<sup>1</sup>, R.C. Mishra<sup>2</sup>, R.P. Tripathi<sup>2</sup>, Arvind K.Srivastava<sup>1\*</sup>, Rolf D. Walter<sup>3</sup>

Divisions of Biochemistry<sup>1</sup> and Medicinal Chemistry<sup>2</sup>,

Central Drug Research Institute, Lucknow (India)

Abteilung für Biochemie, Bernhard Nocht Institute for Tropical Medicine<sup>3</sup>,

Hamburg (Germany)

**Abstract.** A library of 363 glycoconjugates and C-nucleosides synthesized by our earlier reported methods were screened for their effect on isolated filarial glutamate cysteine ligase (GCL) and glutathione reductase (GR). Many of the compounds from the library including one glycosyl urea and one nucleoside showed inhibition of more than 50% on GCL, and seven glycosylated amino esters, two sugars, one glycosyl urea, five glycosyl hydroxamates, one glycosyl hydrantoin, three nucleoside and one N-protected glycopeptide compounds showed inhibition of more than 50% on GR from *S. cervi*. The effect of the most promising ones have been looked on the counterparts from mammalian sources and difference in the susceptibility towards enzyme activity inhibition were noted.

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\*Corresponding author: Tel: +91-522-2212411-17 (Ext. 4346). Fax: +91-522-2223938/ 2223405. E mail: [drarv1955@yahoo.com](mailto:drarv1955@yahoo.com)

## Introduction

Lymphatic filariasis, a disease caused by filarial worms, is a major public health and socioeconomic problem in most of the south-eastern Asian developing countries. In India, an estimated population of 22 million is known to harbour circulating microfilariae and 16 million people suffer from filarial manifestations like elephantiasis of limbs, genitals and hydrocoele. Worldwide about 120 million people are infected with *Wuchereria bancrofti* and *Brugia malayi*. At present there is no effective and safe chemotherapy available against the adult filarial parasites, which raises an urgent need for the development of macrofilaricidal drugs. In the case of filarial worms, which survive for many years in the host, the question has been raised, how they manage this long-term persistence in spite of the host immune response, especially as filarial worms are reported to be sensitive towards oxidative stress.<sup>1</sup> Glutathione (GSH), a tripeptide ( $\gamma$ -glutamylcysteinyl-glycine) is known to guard the filarial worms from oxidative stress and attack by reactive oxygen species; a decrease in glutathione concentration is the most robust and significant alteration in the antioxidant defence.<sup>2</sup> It has been shown that interference with the filarial glutathione metabolism has curative effect on filarial infections in mice.<sup>3</sup> Work so far reveals that parasitic helminths are capable of GSH synthesis; they are known to have very active GSH-linked metabolism; GSH-reductase, GSH-peroxidases and GSH-transferases have been isolated and characterized from a variety of species, and several of these enzymes show chemo- and/or immunotherapeutic potential.<sup>4,5,6</sup> The intracellular concentration of GSH depends on numerous factors but is predominantly controlled by its *de novo* synthesis by glutamate cysteine ligase (GCL; EC 6.3.2.2) and glutathione synthetase, as well as by glutathione reductase (GR, EC 1.8.1.7) which recycles glutathione disulfide back into its reduced and active form.<sup>7,8</sup> Since most normal tissues have a large excess of GSH, and parasites have levels close to that required for survival, the interference with GSH metabolism (GSH synthesis and/ or GSSG reduction) may be more harmful to parasites than to normal tissues of the host. In view of the key role of GCL and GR in dictating GSH availability to filarial worms, inhibition of either one or both may serve as important targets in the design and synthesis of new antifilarial agents. In the present study, GR and GCL have been explored as possible targets for chemotherapeutic attack of this parasitic disease, in accordance with previous reports.<sup>9, 10, 11</sup>

## Experimental

### Chemistry

Synthesis of glycosyl amino esters (11-13, 15-18, acids (14, 19, 20), hydroxamic acids (21-28) and glycopeptides (29-31):

These were prepared following our earlier methods.<sup>13, 14, 15</sup>



Fig 2

Table 2: Glycosyl amino ester acid, hydroxamic acid, N-protected glycopeptide and glycosyl ureids

Compound No.	R	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>
11	CH <sub>3</sub>	2-Propyl-imidazolyl		OCH <sub>2</sub> CH <sub>3</sub>
12 (R)	CH <sub>3</sub>	H	C <sub>7</sub> H <sub>15</sub>	OCH <sub>2</sub> CH <sub>3</sub>
13 (S)	CH <sub>3</sub>	H	C <sub>7</sub> H <sub>15</sub>	OCH <sub>2</sub> CH <sub>3</sub>
14	CH <sub>3</sub>	H	C <sub>7</sub> H <sub>15</sub>	OH
15	CH <sub>3</sub>	H	Furyl	OCH <sub>2</sub> CH <sub>3</sub>
16 (R)	CH <sub>3</sub>	H	CH <sub>2</sub> -2-OH-Ph	OCH <sub>2</sub> CH <sub>3</sub>
17 (S)	CH <sub>3</sub>	H	CH <sub>2</sub> -2-OH-Ph	OCH <sub>2</sub> CH <sub>3</sub>
18	CH <sub>2</sub> Ph	H	Furyl	OCH <sub>2</sub> CH <sub>3</sub>
19	CH <sub>3</sub>	H	3,5-di Me-hexyl	OH
20	CH <sub>3</sub>	H	F-moc	OH
21	CH <sub>3</sub>	H	C <sub>7</sub> H <sub>15</sub>	NHOH
22	CH <sub>3</sub>	H	C <sub>16</sub> H <sub>33</sub>	NHOH
23	CH <sub>2</sub> Ph	H	C <sub>12</sub> H <sub>25</sub>	NHOH
24	CH <sub>2</sub> Ph	H	CH <sub>2</sub> Ph	NHOH
25	CH <sub>2</sub> Ph	H	C <sub>7</sub> H <sub>15</sub>	NHOH
26	CH <sub>2</sub> Ph	H	C <sub>16</sub> H <sub>33</sub>	NHOH
27	CH <sub>2</sub> Ph	H	CH <sub>2</sub> Ph	NHOH
28	CH <sub>2</sub> Ph	Morpholyl		NHOH
29	CH <sub>3</sub>	Fmoc-Proline		OCH <sub>2</sub> CH <sub>3</sub>
30	CH <sub>3</sub>	Proline		OCH <sub>2</sub> CH <sub>3</sub>
31	CH <sub>3</sub>	Valine		OCH <sub>2</sub> CH <sub>3</sub>
32	CH <sub>3</sub>	CH <sub>2</sub> -2-OH-Ph	CONH-3-COCH <sub>3</sub> -Ph	OCH <sub>2</sub> CH <sub>3</sub>
33	CH <sub>3</sub>	CH <sub>2</sub> -2-OH-Ph	CONHCH <sub>2</sub> Ph	OCH <sub>2</sub> CH <sub>3</sub>

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34	CH <sub>2</sub> Ph	H	CONHPh	OCH <sub>2</sub> CH <sub>3</sub>
35	CH <sub>2</sub> Ph	CH <sub>2</sub> Ph	CONHCH <sub>2</sub> Ph	OCH <sub>2</sub> CH <sub>3</sub>
36	CH <sub>3</sub>	3,5-di Me-hexyl	CONHCH <sub>2</sub> Ph	OCH <sub>2</sub> CH <sub>3</sub>
37	CH <sub>2</sub> Ph	Furyl	CONH-4-Cl-Ph	OCH <sub>2</sub> CH <sub>3</sub>
38	CH <sub>3</sub>	n-butyl	C=SNH-3-Cl-Ph	OCH <sub>2</sub> CH <sub>3</sub>
39	CH <sub>3</sub>	n-butyl	C=SNH-2-Cl-Ph	OCH <sub>2</sub> CH <sub>3</sub>
40L	CH <sub>3</sub>	3-O-Me Sugar	CONH-3-Cl-4-Me-Ph	NH <sub>2</sub>
41L	CH <sub>3</sub>	3-O-Me Sugar	CONH-3-COCH <sub>3</sub> -Ph	NH <sub>2</sub>
42L	CH <sub>3</sub>	3-O-Me sugar	CONHCH <sub>2</sub> Ph	NH <sub>2</sub>
43L	CH <sub>3</sub>	3-O-Benzyl sugar	CONH-2-Me-Ph	NH <sub>2</sub>
44L	CH <sub>3</sub>	3-O-Benzyl sugar	CONH-4-Me-Ph	NH <sub>2</sub>
45L	CH <sub>3</sub>	3-O-Benzyl sugar	CONH-3-Cl-4-Me-Ph	NH <sub>2</sub>
46L	CH <sub>3</sub>	3-O-Benzyl sugar	CONH-3-COCH <sub>3</sub> -Ph	NH <sub>2</sub>
47L	CH <sub>3</sub>	CH <sub>2</sub> Ph	CONH-4-Me-Ph	NH <sub>2</sub>
48L	CH <sub>3</sub>	CH <sub>2</sub> Ph	CONH-3-Cl-4-Me-Ph	NH <sub>2</sub>
49L	CH <sub>3</sub>	CH <sub>2</sub> Ph	CONH-3-COCH <sub>3</sub> -Ph	NH <sub>2</sub>
50L	CH <sub>3</sub>	CH <sub>2</sub> Ph	CONHCH <sub>2</sub> Ph	NH <sub>2</sub>
51L	CH <sub>3</sub>	Furyl	CONH-4-Me-Ph	NH <sub>2</sub>
52L	CH <sub>3</sub>	Furyl	CONHCH <sub>2</sub> Ph	NH <sub>2</sub>
53L	CH <sub>3</sub>	Furyl	CONH-2-Me-Ph	NH <sub>2</sub>
54L	CH <sub>3</sub>	CH <sub>2</sub> -4-OCH <sub>3</sub> -Ph	CONH-3-Cl-4-Me-Ph	NH <sub>2</sub>
55L	CH <sub>3</sub>	CH <sub>2</sub> -4-OCH <sub>3</sub> -Ph	CONHCH <sub>2</sub> Ph	NH <sub>2</sub>
56L	CH <sub>3</sub>	CH <sub>2</sub> -2-OH-Ph	CONH-4-Me-Ph	NH <sub>2</sub>
57L	CH <sub>3</sub>	CH <sub>2</sub> -2-OH-Ph	CONH-3-Cl-4-Me-Ph	NH <sub>2</sub>
58L	CH <sub>3</sub>	CH <sub>2</sub> -2-OH-Ph	CONH-3-COCH <sub>3</sub> -Ph	NH <sub>2</sub>
59L	CH <sub>3</sub>	H	CONHCH <sub>2</sub> Ph	NH <sub>2</sub>
60L	CH <sub>3</sub>	H	CONH-2-Me-Ph	NH <sub>2</sub>
61L	CH <sub>3</sub>	H	CONH-4-Me-Ph	NH <sub>2</sub>
62L	CH <sub>3</sub>	H	CONH-3-Cl-4-Me-Ph	NH <sub>2</sub>
63L	CH <sub>3</sub>	H	CONH-4-Me-Ph	NH <sub>2</sub>
64L	CH <sub>3</sub>	CH <sub>2</sub> -4-OCH <sub>3</sub> -Ph	H	NH <sub>2</sub>

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### Synthesis of glycosyl ureas (32-39) and nucleosides (89-99)

Sugar derived N-substituted and N-unsubstituted amino esters were reacted with substituted phenyl isocyanate/isothiocyanate in anhydrous DCM to give N-glycosyl N-substituted urea/thiourea respectively, and the method has been patented both on solid phase as well as by conventional method of synthesis and the chemistry of nucleosides has been published by us.<sup>16,17,18,19,20</sup>

### Solid phase synthesis of glycosyl ureas (40L-64L):

These compounds were synthesized as a library on sieber amide resin in an advanced Chemtech robotic synthesizer in vacuo. Fluorenylmethoxycarbonyl-3-amino- [(1R,

2*R*, 3*S*, 4*R*) -3-*O*-methyl-1, 2-*O*-isopropylidene-1, 4-tetrahydrofuranos-4-yl)-propanoic acid <sup>19</sup> was loaded onto the resin following the standard peptide bond coupling methods (DIC, HOBT and DMAP combination) procedure. The Fmoc protection was removed using 20 % piperidine: DMF. This was followed by reductive alkylation using aromatic and sugar aldehydes in anhydrous trimethyl orthoformate followed by addition of a suspension of NaCNBH<sub>3</sub>. These resin bound compounds were again reacted with different isocyanates in anhydrous DCM to give resin bound glycosyl ureas. Finally, the compounds were cleaved using 2% TFA/DCM and collected in cleavage vials. Compounds of the library, thus obtained, were lyophilized by dissolving in *t*-Butanol/ water (4:1) and drying under freeze gave the title compounds.

**(1*R*, 2*R*, 3*S*, 4*R*, 5*S*)- 5,6-dideoxy-1, 2-*O*-isopropylidene- 3-*O*-methyl-5-*N*<sup>1</sup>-2'-hydroxy benzyl {*N*<sup>3</sup>-4'-chloro phenyl-1-ureidyl}-1,4-heptofuranos-5-yl -uronamide [56L]:**

MS (FAB): 500 (M+H)<sup>+</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 200 MHz): δ 7.30, 7.16 and 7.05 (m, 10H, Ar-H and NH), 7.02 and 6.34 (bs, 2H, D<sub>2</sub>O exchangeable NH), 5.82 (d, *J* = 3.3 Hz, 1H, H-1), 4.55 (d, *J* = 3.3 Hz, 1H, H-2), 4.30-4.19 (m, 4H, H-4, H-5 and NCH<sub>2</sub>), 3.67 (d, *J* = 2.8 Hz, 1H, H-3), 3.39 (s, 3H, -OCH<sub>3</sub>), 3.28 (bs, 1H, NH), 2.60 (m, 2H, H-6), 1.45 and 1.28 [each s, each 3H, C(CH<sub>3</sub>)<sub>2</sub>].

**Solid phase synthesis of glycopeptidyl ureas (65L-88L):**

The synthetic strategy starts with, *N*-Fmoc-3-amino-[(1*R*,2*R*,3*S*,4*R*)-3-*O*-methyl (benzyl)-1,2-*O*-isopropylidene-1,4-tetrahydrofuranos-4-yl]-propanoic acids were anchored on to the activated sieber amide resin using standard protocol. The deprotection of F-moc group of resin bound sugar amino acid was carried out with 30 % piperidine in DMF to give free amine on amino acyl resin. Further coupling of *N*-F-moc α-amino acids was done using the previous method to get resin bound *N*-F-moc glycopeptides compound. Similar deprotection of the F-moc group resulted in free amine functionality. This was followed by sequential reaction with different aldehydes and sodium cyanoborohydride (reductive alkylation) separately resulted in the resin bound glycopeptids. The separate reaction of this resin bound compound

with different substituted aromatic isocyanates resulted in formation of resin bound glycopeptidyl ureids.

Finally the compounds were cleaved off the resin using 2% TFA/DCM cleavage reagent to get glycopeptidyl ureids (15L-16L). Compounds, thus obtained were lyophilized by dissolving them in a mixture of *t*-BuOH and water (4:1) followed by drying under freeze.

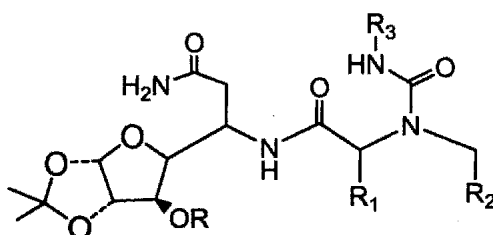


Fig 3: Core structure of glycopeptidyl urea

Table 3: Glycopeptidyl ureids synthesized on solid phase

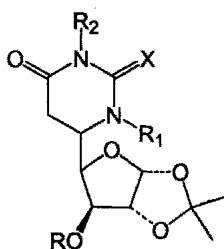
Compound No.	R	R <sub>1</sub> Residue	R <sub>2</sub>	R <sub>3</sub>
65L	CH <sub>3</sub>	Asparagine	CH <sub>2</sub> -4-Br-Ph	4-Cl-Ph
66L	CH <sub>3</sub>	Glutamine	CH <sub>2</sub> Ph	<i>o</i> -Tolyl
67L	CH <sub>3</sub>	Glutamine	CH <sub>2</sub> -4-Br-Ph	4-Cl-Ph
68L	CH <sub>3</sub>	Isoleucine	CH <sub>2</sub> -4-OCH <sub>3</sub> -Ph	CH <sub>2</sub> Ph
69L	CH <sub>3</sub>	Isoleucine	CH <sub>2</sub> Ph	<i>o</i> -Tolyl
70L	CH <sub>3</sub>	Methionine	CH <sub>2</sub> -4-OCH <sub>3</sub> -Ph	CH <sub>2</sub> Ph
71L	CH <sub>3</sub>	Phenyl alanine	CH <sub>2</sub> -4-OCH <sub>3</sub> -Ph	CH <sub>2</sub> Ph
72L	CH <sub>3</sub>	Phenyl alanine	CH <sub>2</sub> Ph	<i>o</i> -Tolyl
73L	CH <sub>3</sub>	Proline	CH <sub>2</sub> -4-Br-Ph	4-Cl-Ph
74L	CH <sub>3</sub>	Valine	CH <sub>2</sub> -4-Br-Ph	4-Cl-Ph
75L	CH <sub>2</sub> Ph	Isoleucine	CH <sub>2</sub> -4-OCH <sub>3</sub> -Ph	CH <sub>2</sub> Ph
76L	CH <sub>2</sub> Ph	Isoleucine	CH <sub>2</sub> Ph	<i>o</i> -Tolyl
77L	CH <sub>2</sub> Ph	Isoleucine	Furyl	<i>p</i> -Tolyl
78L	CH <sub>2</sub> Ph	Isoleucine	CH <sub>2</sub> -4-Br-Ph	4-Cl-Ph

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79L	CH <sub>2</sub> Ph	Leucine	CH <sub>2</sub> Ph	<i>o</i> -Tolyl
80L	CH <sub>2</sub> Ph	Leucine	CH <sub>2</sub> -4-Br-Ph	4-Cl-Ph
81L	CH <sub>2</sub> Ph	Methionine	CH <sub>2</sub> -4-OCH <sub>3</sub> -Ph	CH <sub>2</sub> Ph
82L	CH <sub>2</sub> Ph	Methionine	Furyl	<i>p</i> -Tolyl
83L	CH <sub>2</sub> Ph	Methionine	CH <sub>2</sub> -4-Br-Ph	4-Cl-Ph
84L	CH <sub>2</sub> Ph	Phenyl alanine	CH <sub>2</sub> -4-OCH <sub>3</sub> -Ph	CH <sub>2</sub> Ph
85L	CH <sub>2</sub> Ph	Phenyl alanine	CH <sub>2</sub> Ph	<i>o</i> -Tolyl
86L	CH <sub>2</sub> Ph	Proline	CH <sub>2</sub> -4-Br-Ph	4-Cl-Ph
87L	CH <sub>2</sub> Ph	Valine	CH <sub>2</sub> Ph	<i>o</i> -Tolyl
88L	CH <sub>2</sub> Ph	Valine	Furyl	<i>p</i> -Tolyl

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**Solid phase synthesis of C-nucleosides (100L-119L):** C-nucleosides were synthesized as reported in our previous paper.<sup>21</sup>



**Fig 4:** Core structure of Nucleosides

**Table 4:** C-Nucleosides synthesized by conventional method and on solid phase

Compound No.	R	R <sub>1</sub>	R <sub>2</sub>	X
89	CH <sub>3</sub>	Furyl	4-n-Bu-Ph	O
90	CH <sub>3</sub>	CH <sub>2</sub> -2-OH-Ph	3-COCH <sub>3</sub> -Ph	O
91	CH <sub>2</sub> Ph	H	CH <sub>2</sub> Ph	O
92	CH <sub>3</sub>	C <sub>12</sub> H <sub>25</sub>	CH <sub>2</sub> Ph	O
93	CH <sub>3</sub>	3,5-di Me-hexyl	CH <sub>2</sub> Ph	O
94 (S)	CH <sub>2</sub> Ph	CH <sub>2</sub> Ph	CH <sub>2</sub> Ph	O
95	CH <sub>2</sub> Ph	3,5-di Me-hexyl	4-F-Ph	O

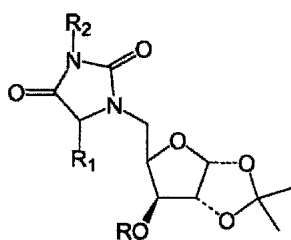
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96 (R)	CH <sub>2</sub> Ph	CH <sub>2</sub> Ph	CH <sub>2</sub> Ph	O
97	CH <sub>2</sub> Ph	Furyl	4-Cl-Ph	O
98	CH <sub>3</sub>	n-butyl	3-Cl-Ph	S
99	CH <sub>3</sub>	n-butyl	2-Cl-Ph	S
100L	CH <sub>3</sub>	CH <sub>2</sub> Ph	2,4-di-Cl-Ph	O
101L	CH <sub>3</sub>	CH <sub>2</sub> Ph	4-n-Bu-Ph	O
102L	CH <sub>3</sub>	3-OH,4-OMe-Ph	3,4-di-Cl-Ph	O
103L	CH <sub>3</sub>	3-OH,4-OMe-Ph	3,5-di-Cl-Ph	O
104L	CH <sub>3</sub>	Furyl	4-n-Bu-Ph	O
105L	CH <sub>2</sub> Ph	CH <sub>2</sub> -4-OCH <sub>3</sub> -Ph	4-Acetyl Ph	O
106L	CH <sub>3</sub>	CH <sub>2</sub> -4-OCH <sub>3</sub> -Ph	3-Acetyl Ph	O
107L	CH <sub>3</sub>	CH <sub>2</sub> Ph	3,5-di-Me-Ph	O
108L	CH <sub>3</sub>	CH <sub>2</sub> Ph	CH <sub>2</sub> Ph	O
109L	CH <sub>3</sub>	CH <sub>2</sub> Ph	3-Cl-4-Me-Ph	O
110L	CH <sub>3</sub>	CH <sub>2</sub> Ph	3,5-di-Me-Ph	O
111L	CH <sub>3</sub>	Furyl	3-Acetyl Ph	O
112L	CH <sub>3</sub>	Furyl	4-CN-Ph	O
113L	CH <sub>3</sub>	CH <sub>2</sub> -2-OH-Ph	3-Acetyl Ph	O
114L	CH <sub>3</sub>	CH <sub>2</sub> -2-OH-Ph	CH <sub>2</sub> Ph	O
115L	CH <sub>3</sub>	CH <sub>2</sub> -2-OH-Ph	3-Cl-4-Me-Ph	O
116L	CH <sub>3</sub>	CH <sub>2</sub> -2-OH-Ph	4-CN-Ph	O
117L	CH <sub>2</sub> Ph	H	4-Cl-Ph	O
118L	CH <sub>3</sub>	H	4-Cl-Ph	O
119L	CH <sub>2</sub> Ph	H	4-Acetyl Ph	O

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However, during our synthetic strategy for the synthesis of glycosyl urea, when the glucose derived aldehyde was reacted with leucine methyl ester hydrochloride in presence of triethyl amine and sodium borohydride followed by the reaction of the intermediate with 3-acetylphenyl isocyanate resulted in the corresponding cyclised urea which on refluxing in presence of DBU and TBAB gave a five-membered glycosyl hydrantoin (compound 120) in good yield.





**Fig 5: Structure of compound 120**

R=CH<sub>2</sub>Ph, R<sub>1</sub>= Leucine, R<sub>2</sub> = 3-Acetyl Ph

The synthesis of compounds 121 and 122 has already been reported in our earlier paper.<sup>22</sup>

### Biological

**Chemicals:** Tris, phosphoenol pyruvate, sodium L-glutamate, L- $\alpha$ -aminobutyric acid, sodium EDTA,  $\beta$ -nicotinamide adenine dinucleotide (reduced), adenosine-triphosphate, L-buthionine-[S,R]-sulfoximine (BSO), cystamine, carmustine and methylene blue were purchased from Sigma Chemical Co., USA. Potassium chloride, magnesium chloride, pyruvate kinase, lactate dehydrogenase, oxidised glutathione and  $\beta$ -nicotinamide adenine dinucleotide phosphate (reduced) were purchased from Sisco Research Laboratories Pvt. Ltd Mumbai, India. All other chemicals used were of analytical grade.

**Biological Material:** Adult *Setaria cervi* females of average body weight 35 $\pm$ 5 mg and length 6.0 $\pm$ 1.0 cm were collected from the peritoneal cavity of freshly slaughtered naturally infected water buffaloes *Bubalus bubalis* (Linn.) at a local abattoir. They were thoroughly washed with saline and kept in Hanks balanced salt solution with sodium bicarbonate and glucose (5.55 mM) for 60 min for complete revival before being used in enzymic studies.

## **Methods**

### **Preparation of GCL and GR from bovine filarial worms *S. cervi***

A 10% homogenate of actively motile bovine filarial worms was prepared in a Tris-HCl buffer, 50 mM, pH 7.5 containing 1mM glutamate, 5 mM MgCl<sub>2</sub> and 0.1 mM PMSF using a Potter Elvehjem glass homogenizer fitted with a Teflon pestle. The homogenate was centrifuged at 1,000 x g for 15 min, 10,000 x g for 30 min, and subsequently at 100,000 x g for 60 min to obtain mitochondrial, post mitochondrial and cytosolic fractions, respectively. The cytosolic fraction was then subjected to ammonium sulfate precipitation. For the goal, the fraction was brought to a final concentration of 0-45% saline saturation with solid ammonium sulfate. The sample was stirred in cold for 30 min and then centrifuged at 13,000 x g for 15 min to collect the precipitate and the supernatant separately. The supernatant solution was then saturated with solid ammonium sulfate to a concentration of 65%. The sample was again stirred for 30 min and centrifuged at 13,000 x g for 15 min to collect the precipitate and the supernatant separately. The supernatant solution was then saturated with solid ammonium sulfate to a final concentration of 80%. After stirring in cold for 30 min, the suspension was centrifuged at 13,000 x g for 15 min. All the pellets were dissolved separately in the homogenisation buffer. All the samples were analysed for GCL and GR activities, which were found enriched in the 65-80% ammonium sulfate precipitate. This fraction was stored at 4°C and used as the source of GCL and GR.

### **GCL Activity determination**

GCL activity was estimated in the filarial extracts following a coupled enzyme procedure in which the rate of ADP formation in the presence of pyruvate kinase, LDH, PEP and NADH is obtained from the decrease in absorbance of NADH at 340 nm.<sup>23</sup> Enzyme activity was monitored at 37° C using a Shimadzu double beam UV spectrophotometer in reaction mixtures (final volume 1.0 ml) containing 100 mM Tris HCl buffer; pH 8.4, 150 mM KCl, 5 mM ATP, 2 mM PEP, 10 mM L-glutamate, 10 mM L- $\alpha$ -aminobutyrate, 20 mM MgCl<sub>2</sub>, 2 mM Na<sub>2</sub>EDTA, 0.2 mM NADH, 1.6  $\mu$ g/ $\mu$ l pyruvate kinase and 1.6  $\mu$ g/ $\mu$ l lactate dehydrogenase. Reaction was initiated by the addition of enzyme protein. The absorbance at 340 nm was monitored for 5 min at 30 sec intervals.

### **GR Activity determination**

GR activity in filarial extracts was determined by the method described by Carlberg and Mannervik.<sup>24</sup> All enzyme kinetics experiments were carried out at 37°C using a Shimadzu double beam UV spectrophotometer. Enzyme activity was spectrophotometrically monitored from the rate of conversion of NADPH to NADP ( $\epsilon_{340} = 6.2 \text{ mM}^{-1} \text{ cm}^{-1}$ ). For each assay the absorbance of NADPH in the reaction mixture was measured over a period of 5 min. Assays were carried out in 1 ml of the following solution: 100 mM potassium phosphate buffer, pH 7.0; 1mM EDTA; 1 mM GSSG; 0.2 mM NADPH and enzyme protein. Enzyme aliquots were preincubated in the presence of NADPH for 5 min in order to trace the non-enzymatic oxidation, after which GSSG was added to start the reaction. To correct for non-specific oxidation of NADPH, the blank cuvette contained all the assay components except GSSG.

Protein was estimated following the method of Lowry *et al.*<sup>25</sup> A unit of enzyme activity was expressed as the amount that catalyzes the consumption of 1 $\mu$ mol of substrate per minute, using a molar extinction coefficient of  $6.2 \times 10^6$  for NADH/NADPH.

### **Inhibition studies**

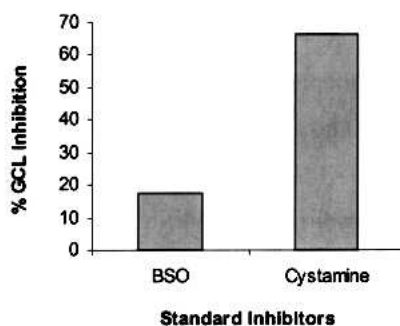
Effect of standard inhibitors (BSO, cystamine, methylene blue and carmustine) and compounds belonging to various chemical series on GCL and GR activity of *S. cervi* females was studied by adding them directly in the assay systems at 200  $\mu$ M concentration 10 min prior to addition of the substrate. The percentage inhibition/stimulation of the enzyme activity by compounds were calculated by comparing with control tube.

### **Results and Discussion**

Bovine filarial worms, *S. cervi*, were found equipped with both the GSH synthetic as well as GSSG reductive pathways and the specific activities of the GCL and GR in crude filarial extracts were calculated to be around 0.12 - 0.15 and 0.8 - 1.0  $\mu$ mol/min/mg protein, respectively.

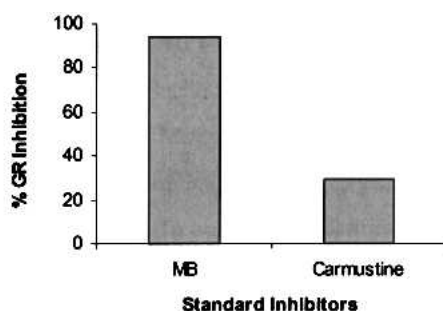
Figs 6 and 7 show the pattern of inhibition of standard mammalian inhibitors i.e., buthionine sulfoximine and cystamine on filarial GCL and menadione and carmustine

on GR at 200  $\mu\text{M}$  concentration. Under the specified assay conditions the maximum inhibition of BSO was found to be 16.1% at 200  $\mu\text{M}$  concentration which indicates that BSO is a weak inhibitor of filarial GCL. It inhibited GCL in confirmation of most<sup>26</sup> but not of all living cells.<sup>27</sup> Many of the published reports also indicate that BSO is a relatively poor inhibitor of bacterial GCL, particularly in the presence of physiological levels of L-glutamate. In contrast, cystamine shows strong inhibition on GCL from *S. cervi*, even at low concentrations. *S. cervi* GCL was inhibited by cystamine to the tune of 19-67% when studied in the concentration range of 0.25-200  $\mu\text{M}$ . It has previously been found that the disulfide cystamine (2, 2'-dithiobis (ethylamine)) is a potent inactivator of GCL, and the susceptibility of this enzyme to inactivation is markedly influenced by the presence of magnesium ions, substrates and products.<sup>28</sup> Carmustine (1,3-bis[2-chloroethyl]-1-nitrosourea) (BCNU), a potent inhibitor of GR,<sup>29</sup> inhibited *S. cervi* GR to the tune of 20-58.5% when tested in 10 - 200  $\mu\text{M}$  range. Methylene blue,<sup>30</sup> known mammalian GR inhibitor also showed appreciable inhibition to the tune of 24-94% when tested in the 2 - 200  $\mu\text{M}$  range on *S. cervi* GR.



**Fig 6:** Effect of standard inhibitors on *S. cervi* GCL

\* Concentration of inhibitors in the assay mixture- 200 $\mu\text{M}$



**Fig 7:** Effect of standard inhibitors on *S. cervi* GR

\* Concentration of inhibitors in the assay mixture- 200 $\mu$ M

In an effort to search for novel inhibitors of filarial GCL and GR, many of the synthetic compounds that may serve as substrate analogs for GSH biosynthesis and hamper its *de novo* synthesis, belonging to various chemical series i.e. glycosylated amino acids (GAA), glycosylated amino esters (GAE), sugars, diglycosylated amino esters, glycosylated sulfone derivatives, thioglycosylated esters, Fmoc derivatives of GAE, glycosyl ureas and thioureas, oxime, glycosylated hydroxamic acid, glycosylated hydrantoin, glycopeptides, nucleosides, C-nucleosides and S-containing nucleoside, thiazidine thiones, N-protected glycopeptide were looked for their *in vitro* effect at 200  $\mu$ M concentration in the assay systems. Table 5 represents the comparative effects of selective synthetic compounds and standard mammalian inhibitors on filarial and mammalian counterparts.

Of the various compounds tested for inhibitory effect on filarial GCL and GR, eighty one compounds inhibited GCL to the tune of 10 - 82% of which one glycosyl urea (compound 49L) and one nucleoside (compound 108L) showed more than 50% inhibition on intact GCL, while ninety five compounds inhibited GR to the tune of 10-92% of which, seven glycosylated amino esters (compound 11, 13(s), 14, 15, 16(R), 17(S) and 18), two sugars (compound 121, 122), one glycosyl urea (compound 36), five glycosyl hydroxamates (compound 22, 23, 24, 25, 28), one glycosyl hydrantoin (compound 120), three nucleosides (compound 93, 94, 105L) and one N-protected glycopeptide (compound 29) compounds showed more than 50% inhibition on GR from *S. cervi*. Noteworthy in this connection are fifteen compounds that showed

simultaneous inhibition on GCL and GR. The compounds which showed more than 50% inhibition on GCL/ GR were screened on the counterparts from mammalian sources for the optimization of the leads. Majority of the compounds showed no inhibitory activity on the mammalian enzymes except glycosyl urea (compound 49L) which showed 18.0% inhibition on mammalian GCL, and glycosyl hydroxamate (compound 22), which showed 71.4% inhibition on mammalian GR.

**Table 5:** Comparative susceptibility of filarial and mammalian GCL and GR towards synthetic compounds and standard inhibitors

Compound No.	% GCL Inhibition		% GR Inhibition	
	<i>S. cervi</i>	Mice liver	<i>S. cervi</i>	Mice liver
11	32	ND	69	NIL
12 (R)	NIL	ND	42	ND
13 (S)	NIL	ND	77	NIL
14	NIL	ND	67	NIL
15	27	ND	67	NIL
16 (R)	27	ND	55	NIL
17 (S)	49	ND	67	NIL
18	20	ND	67	NIL
19	34	ND	55	NIL
20	8	ND	42	ND
21	4	ND	44	ND
22	NIL	ND	75	71
23	22	ND	50	NIL
24	NIL	ND	56	NIL
25	NIL	ND	54	NIL
26	17	ND	NIL	ND
27	13	ND	NIL	ND
28	23	ND	69	NIL
29	13	ND	92	NIL
30	22	ND	NIL	ND
31	21	ND	NIL	ND
32	NIL	ND	+29	ND
33	NIL	ND	+3	ND
34	NIL	ND	11	ND
35	25	ND	NA	ND
36	NIL	ND	68	NIL
37	21	ND	NIL	ND
38	28	ND	NIL	ND
39	26	ND	NIL	ND
40L	NIL	ND	21	ND
41L	NIL	ND	18	ND
42L	NIL	ND	29	ND
43L	NIL	ND	32	ND
44L	NIL	ND	29	ND
45L	NIL	ND	16	ND
46L	NIL	ND	19	ND

(Continued next page.)

47L	NIL	ND	16	ND
48L	NIL	ND	18	ND
49L	82	18	14	ND
50L	NIL	ND	33	ND
51L	NIL	ND	16	ND
52L	NIL	ND	38	ND
53L	NIL	ND	25	ND
54L	NIL	ND	35	ND
55L	NIL	ND	19	ND
56L	NIL	ND	43	ND
57L	NIL	ND	21	ND
58L	NIL	ND	27	ND
59L	NIL	ND	27	ND
60L	NIL	ND	16	ND
61L	16	ND	29	ND
62L	14	ND	NA	ND
63L	17	ND	NA	ND
64L	NIL	ND	30	ND
65L	34	ND	NA	ND
66L	NIL	ND	25	ND
67L	21	ND	NA	ND
68L	NIL	ND	18	ND
69L	NIL	ND	17	ND
70L	+34	ND	NA	ND
71L	NIL	ND	40	ND
72L	NIL	ND	25	ND
73L	NIL	ND	29	ND
74L	NIL	ND	27	ND
75L	29	ND	NA	ND
76L	27	ND	NA	ND
77L	15	ND	NA	ND
78L	32	ND	NA	ND
79L	19	ND	NA	ND
80L	40	ND	NA	ND
81L	25	ND	NA	ND
82L	28	ND	NA	ND
83L	24	ND	NA	ND
84L	18	ND	NA	ND
85L	30	ND	NA	ND
86L	NIL	ND	35	ND
87L	NIL	ND	38	ND
88L	27	ND	33	ND
89	22	ND	NIL	ND
90	NIL	ND	46	ND
91	NIL	ND	+86	ND
92	NIL	ND	36	ND
93	NIL	ND	75	NIL
94	NIL	ND	72	NIL
95	37	ND	NIL	ND
96	21	ND	NA	ND
97	17	ND	NIL	ND
98	10	ND	NIL	ND
99	19	ND	40	ND

(Continued next page.)

100L	+21	ND	NA	ND
101L	+25	ND	NIL	ND
102L	NIL	ND	6	ND
103L	NIL	ND	49	ND
104L	+20	ND	NIL	ND
105L	NIL	ND	60	NIL
106L	+36	ND	8	ND
107L	NIL	ND	19	ND
108L	57	NIL	28	ND
109L	NIL	ND	28	ND
110L	NIL	ND	26	ND
111L	25	ND	19	ND
112L	NIL	ND	19	ND
113L	NIL	ND	43	ND
114L	+22	ND	NIL	ND
115L	NIL	ND	26	ND
116L	NIL	ND	34	ND
117L	+23	ND	26	ND
118L	NIL	ND	17	ND
119L	NIL	ND	15	ND
120	25	ND	88	NIL
121	18	ND	52	NIL
122	13	ND	71	NIL

(Continued from previous page.)

\*Concentration of test sample in the assay mixture-200µM

NIL= No inhibition; NA= Not available; ND= Not done

The antiparasitic agents known till date, act either on the membrane receptors or on enzyme/metabolic pathways crucial for the survival of the parasites.<sup>31</sup> Both available filaricides DEC and ivermectin are able to eliminate microfilaraemia in human lymphatic filariasis, however, the adult parasites mostly remain unaffected resulting in relapse of microfilaraemia.<sup>32</sup> Therefore, there is an urgent need to develop an antifilarial drug which can kill the adult parasites slowly, has microfilaricidal activity and at the same times sterilize the female worms. GSH dependent enzymes and GSH itself have a key role in protecting tissues against oxidative stress caused by free radicals and their reaction products. In helminths, this is of particular interest with respect to protection against respiratory burst of activated cells of the host's immune system and susceptibility to drugs whose mode of action involves the generation of free radicals.<sup>33</sup> In view of the high probability that the maintenance of an optimal GSH/GSSG ratio is vital to filariae, it seems worthwhile to develop drugs that could selectively deplete or distort the glutathione stores in these parasites.



In an approach towards this end, the present study was designed to try a preliminary rating of target attractiveness within the area of GSH-dependent parasitic antioxidant defence. The proposed compounds in the present study are expected to inhibit two crucial enzymes, GCL and GR, which are essential for the survival and propagation of these parasites.

It has been shown that certain sugar derivatives possess very good activity against filarial glutathione metabolizing enzymes *in vitro*.<sup>12,34,35,36</sup> These compounds possess immunomodulatory activity<sup>37</sup> and *in vitro* antifilarial activity too. As sugars are known to offer better pharmacokinetics, better transport and above all less toxicity<sup>38,39</sup> to the drugs, it is presumed that amino acid analogues if grafted on sugar backbone may serve as substrate for GSH biosynthesis and may hamper its *de novo* synthesis. C-nucleosides having C-C linkage instead of carbon to nitrogen linkage between the aglycon and the sugar moiety are known mainly for their anticancer,<sup>39</sup> antiviral<sup>40</sup> and antileukemic<sup>41</sup> activities. A C-C bond between the sugar and aglycon makes them more stable towards enzymatic degradation, and therefore the half-lives of these compounds in the body are far greater than the corresponding N-nucleosides. In addition, sugar derivatives are known to offer better stability, better pharmacokinetic parameters, facilitate the transport of drugs, and at the same time are less toxic.

If the present strategy i.e. inhibition of parasite GCL/GR activity, affects the survival of the parasites or helps in the enhancement of activity of presently available antifilarial/antimalarial drugs, it will help in the design of new target based chemotherapeutic agents. The compounds which, in the present study, inhibit GCL/GR more than 50% and those which show simultaneous inhibitory effect on GCL and GR can thus emerge as lead molecules for further exploration to develop new class of antifilarial drugs.

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